

L 14419-63

ACCESSION NR: AP3004082

titanate having a wide particle-size distribution confirmed the hypothesis that strontium oxide is leached out of the titanate with methanol. Charge reversal occurred after 14-15 days of uninterrupted flow of methanol through the strontium titanate diaphragm; the titanium dioxide diaphragm remained negatively charged. The zeta potential dropped within 20 days from +71 to -8 mv in the strontium titanate system, but remained constant at -11 mv in the titanium dioxide system. The observed decrease in electrical conductivity of methanol in the pores of the strontium titanate diaphragm cannot by itself lead to charge reversal. "The authors are grateful to S. P. Tibilov for suggesting the subject of research." Orig. art. has: 7 figures and 2 tables.

ASSOCIATION: none

SUBMITTED: 23Apr63

DATE ACQ: 15Aug63

ENCL: 01

SUB CODE: PH, CH

NO REP Sov: 005

OTHER: 001

Cord 3/4

L 14419-63
ACCESSION NR: AP3004082

ENCLOSURE: 61

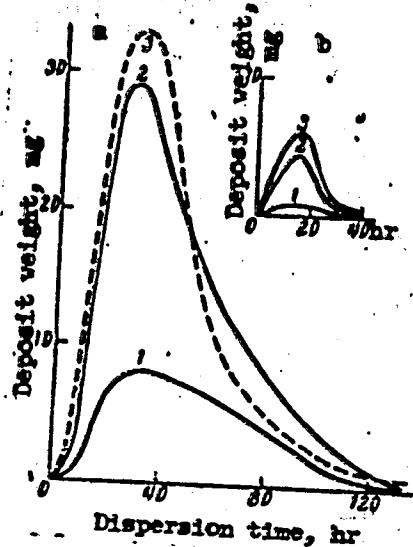


Fig. 1. Deposit weight versus comminution time

Strontium titanate concentration: 5 g.(a)
and 1 g.(b) per 100 ml methanol. Potential
gradient: 1 - 5 v/cm; 2 - 15 v/cm;
3 - 25 v/cm.

Card 4/4

L 10755-63

EWP(q)/EWT(m)/BDS—AFFTC/ASD—JD

ACCESSION NR: AP3002894

S/0054/63/000/002/0101/0107

AUTHOR: Ostrovskiy, V. V.; Styrkas, A. D.57
55TITLE: Electrolytic deposition of indium antimonideSOURCE: Leningrad. Universitet. Vestnik. Seriya fiziki i khimii, no. 2, 1963,
101-107

TOPIC TAGS: indium antimonide, indium, antimony, intermetallic compounds, single crystals, films, electrolytic deposition, single deposition, cathodic potential, electrolytic deposition, cathodic polarization, limiting current, indium sulfate, antimony sulfate, Hall pickup, infrared detector

ABSTRACT: The growth of cathodic films of indium antimonide (InSb) single crystals by electrolytic deposition from aqueous antimony and indium sulfate solutions has been investigated. The study was prompted by the apparent simplicity of cathodic deposition which can produce a film of InSb single crystals of a desired thickness. It is noted that the semiconductor properties

Card 1/13

L 10752-63
ACCESSION NR: AP3002894

of InSb single crystals make them suitable for use in Hall pickups and infrared detectors. Cathodic deposition processes for In and Sb separately were studied as a preliminary step to the investigation of cathodic codeposition. Cathodic potentials were measured by the opposition method in electrolytes of varied composition and pH in a diaphragm cell under a carbon dioxide atmosphere. The Sb content of the cathodic deposit was determined by amperometric titration. Experimental cathodic polarization curves for Sb, In, and Sb + In deposits were plotted for solutions of the respective sulfates at various pH values and with or without agitation. It was established that 1) in the absence of In, compact crystalline Sb deposits with 96% current output can be obtained at 0.5 mamp/cm² from electrolytes with pH 0.45; 2) in the absence of Sb, comparable results can be achieved in In solutions within the 1-5 mamp/cm² current-density range at pH 2.5; 3) in solutions containing both Sb and In, the cathodic potential at the limiting current for Sb falls 200 mv short of the indium sulfate decomposition potential. Nevertheless, chemical analysis indicated the possibility of codeposition of In with Sb. However, a deposit with composition

Cord 2/A

L 10755-63
ACCESSION NR: AP3002894

2

corresponding to InSb is difficult to achieve, as suggested by Fig. 1 of the Enclosure. Deposits containing more than 15% Sb are porous and friable. Cathodic-deposits with compositions close to InSb contain Sb, In, and InSb as separate phases, as indicated by x-ray analysis. Subsequent heating of these deposits in a hydrogen atmosphere produced an increase in InSb content at the expense of the Sb and In phases. The high activation energy of formation of InSb from the elements is thought to be the primary obstacle to successful cathodic deposition of InSb. "The authors are grateful to S. P. Tibilov and R. L. Myuller for their assistance in the work." Orig. art. has: 4 figures and 3 tables.

ASSOCIATION: none

SUBMITTED: 17Dec62 DATE ACQ: 24Jul63 ENCL: 01

SUB CODE: 00 NO REF SOV: 013 OTHER: 016

Card 3/13

OSTROUMOV, Viktor Vasil'yevich; IVANOV, S.M., red.; NAZAROVA, A.S.,
tekhn. red.

[Technological innovations form the basis for the development of
industrial production] Tekhnicheskii progress - osnova razvitiia
proizvodstva. Moscow, Izd-vo "Znanie," 1961. 61 p. (Narodnyi
universitet kul'tury: Tekhniko-ekonomicheskii fakul'tet, no.16-17)
(MIRA 15:2)

(Technological innovations) (Industry)

OSTROUMOV, Vasil'yevich; IVANOV, S.M., red.; NAZAROVA, A.S.,
tekhn. red.

[Technological progress is the basis of industrial development]
Tekhnicheskii progress - osnova razvitiia proizvodstva. Mo-
skva, Izd-vo "Znanie," 1961. 47 p. (Vsesoiuznoe obshchestvo
po rasprostraneniiu politicheskikh i nauchnykh znanii. Ser.4,
Tekhnika, no.21) (MIRA 15:2)

(Technology)

OSTROUMOV, V.V. (Leningrad)

Change in some optical properties of palladium under the
influence of hydrogen. Zhur. fiz. khim. 34 no. 11:2571-2578
N '60. (MIRA 14:1)
(Palladium) (Hydrogen)

OSTROUMOV, V.V.; PLOKHOTNIKOVA, I.P.

Deposition of copper from acid electrolytes by means of periodically reversing current. Zhur. prikl. khim. 31 no.10:1520-1526 O '58.

(MIRA 12:1)

(Copy r plating)

OSTROUMOV, V.V.; PLOKHOTNIKOVA, I.F.

Effect of a chlorine ion admixture on the copper plating process
using periodically reversed current. Zhur.prikl.khim. 31 no.11:
1668-1673 N 1958. (MIRA 12:2)

(Copper plating) (Chlorine)

AUTHORS: Ostreumov, V.V., Flokhotnikova, I.F.

TITLE: Some Properties of Electrolytic Copper Precipitation From Acid Electrolytes by Means of Periodic Currents (Nekotorye svoystva elektrolyticheskogo zolotoobrazovaniya v kislykh elektrolitakh s pomosch'yu periodicheskikh obrazchayemykh tokov)

PUBLICAL: Zhurnal prikladnoy khimii, 1969, Vol. XLI(1), p. 17,
pp. 317-320 (USSR)

ABSTRACT: This investigation is a continuation of the work of S. S. Lef. L. P. T. Electrolytic copper is obtained from copper sulfate in the presence of sulfuric acid. If copper is precipitated on a polished glass surface, if i.e. if $i = 10 \text{ mA/cm}^2$ is used, the precipitate has a crystalline structure (Figure 1, a). At a current density the precipitation is of regular order (Figure 1, b). The first layers are oriented particularly to the electrode plane (Figure 1, c). The outer layers precipitated at $i = 10$ have a fibrous structure (Figure 2, b). The sublayers give a determining influence on the structure of the precipitates to a thickness of $\sim 10 \mu$ (Figure 3). The introduction of chlorine ions into the electrolyte at a concentration of $10 - 20 \text{ mg/l}$ restricts the

Card 1/2

S. V. - 5. - 4

Some Properties of Electrolytic Polymers Prepared from Anodic Oxidation of Electrolytes by Means of Periodically Reversing Current

process of anodic polarization formation. The thickness of the layers formed by i_0 of 20 mA/cm² is 1.0 μ m/mm² and rises to 1.5 μ m/mm² at 200 mA/cm². There are two electrochemical paths with a conversion.

SUBMITTER: April 1, 1971

Card 2/2

5(2)

17-10-1984

AUTHOR: STREL'CIK, V.V.

TITLE: Electropolishing rates of Ni are obtained in the presence of Sulfonated Naphthalene. Elektropoliticheskaya tsita v usloviyakh polucheniya nafthalena

PERIODICAL: Zhurnal prikladnoi khimii, XXV, No 7, p. 1517-1520, 'USSR'

ABSTRACT: The presence of sulfonated naphthalene in nickel electrolytes causes irregularities in the surface of the precipitate. If the thickness of the layer reaches 1.4 mm the roughness increases so that protrusions of more than 1 mm rise above the surface. There is a smoothing effect in nickel layers which have not been obtained from electrolytes containing sulfonated naphthalene (Figure 4). Geometrical considerations lead to the conclusion that every precipitated metal smoothes micro roughnesses. The hardness of sulfonated naphthalene precipitated in nickel electrolytes with respect to Cu is higher in comparison to other nickel layers at 400°C. At the start of 400°C of sulfonated naphthalene in the electrolyte the hard layer is 0.15-0.2 mm. It is parabolically decreasing with increasing temperature. The hardness is

Card 1/2

Electrolytic Preparation
Naphthalene

1. Name of Company
2. Address
3. Name of Person
4. Position
5. Name of Person
6. Position
7. Name of Person
8. Position
9. Name of Person
10. Position

SUBMITTED

Card 2 2

OSTROUMOV, V.V.

Influence of some organic substances on the process of palladium plating. Zhur.prikl. khim. 31 no.3:402-408 Mr '58.

(MIRA 11:4)

(Palladium) (Electroplating)

OSTROUMOV, V.V.

Electrodeposition of palladium from solutions, containing alkali hydroxide. Zhur. prikl. khim. 31 no.1:77-83 Ja '58. (MIRA 11:4)
(Electroplating) (Palladium)

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7

OSINOVSKY, V. M.

"A Magnetic Test in Eye Diseases," Vest. Oftal. Ak., No. 1, 1949.

Leningrad Sci. Res. Inst. Eye Diseases Dr. Girsman, -c1949-

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7"

O. TROUMOV, V. . .

Chairman, Sec.

Dissertation: "Investigation of the process of diffusion nitriding of steels
and on strengthening the surface layer during the heat treatment of
alloy steels in the broad temperature range." Moscow

Moscow Major of the Higher Education Institute Doc. Inv. No. 1

1970

SO Vecheryaya Moskva
Sum 71

S/076/60/034/011/016/024
B004/B064

AUTHOR: Ostroumov, V. V. (Leningrad)

TITLE: Change of Some Optical Properties of Palladium Under the Action of Hydrogen

PERIODICAL: Zhurnal fizicheskoy khimii, 1960, Vol 34, No. 11,
pp. 2571-2578

TEXT: The author investigated the reflection of white light on electrolytically produced, 0.1μ thick palladium layers deposited from an amino chloride electrolyte onto the polished surface of a bronze or brass cathode. The light reflection of the highly polished Pd layers was 67%. In a vessel provided with a window, these layers were treated with nitrogen, hydrogen, and a nitrogen-hydrogen mixture, and the reflection was measured with a selenium photocell. The following results were obtained: 1) A substitution of nitrogen for air does not change reflection. 2) A substitution of hydrogen for nitrogen reduces reflection by 25%. This process is accelerated by a temperature increase. 3) In nitrogen-hydrogen mixtures, reflection is reduced in accordance with the partial pressure of hydrogen

Card 1/2

Change of Some Optical Properties of Palladium S/076/60/034/011/016 'CMA
Under the Action of Hydrogen B004/B064

4) If hydrogen (or the nitrogen-hydrogen mixture) is again replaced by air,
the reflection of the Pd layer rises again to the initial value. The same
effect is obtained when the layer is dipped into oxidizing solutions or
polished with an aqueous lime suspension. 5) The reflection is also
reduced in the case of cathodic polarization of Pd in alkaline solutions
or in the joint electrolytic separation of palladium and hydrogen. There
are 8 figures and 7 references: 4 Soviet, 3 US, 1 Austrian, 1 French, and
1 German.

SUBMITTED: March 4, 1959

Card 2/2

OSTROUMOV, YE. A.

USSR:

A method for the determination of different forms of combined sulfur in the sediments of the Black Sea. E. S. Ostroumov. *Trudy Inst. Okeanol., Akad. Nauk S.S.R.* No. 22, 1962. A method for the detn. of 6 types of S on the same sample was developed; this method was based on the fact that HNO_3 , d. 1.2, oxidizes pyritic and not org. S. Treat the dry sample with 6*N* HCl, and det. the H₂S as Cu^{2+} ; treat the HCl soln. with 10% BaCl₂ and det. SO_4^{2-} gravimetrically. Ext. free S from the residue with Me_2CO (cf. Romm, C.A. 39, 11181) and det. it colorimetrically as $Fe(SCN)_3$. Allow the residue contg. pyritic and org. S to stand with 6*N* HNO_3 for 5 days with occasional shaking; filter, and treat the residue with Br⁻-contd. HNO_3 . Treat both solns. with BaCl₂, and analyze as usual.
I. Bencowitz

~~OSTROUMOV~~

OSTROUMOV, Ye-A

USSR;

Different forms of combined sulfur in the sediments of the Black Sea. I. A. Ostroumov. Trudy Inst. Okeanol. Akad. Nauk S.S.R., 70-71 (1983).—Borings up to 250 cm. in depth were taken from the bottom of the Black Sea at 13 positions parallel to the coast line at sea depths up to 2133 m. The percentage of the different types of S in the sediment was plotted as a function of the depth of profile. The proportion of FeS and elemental S decreased with the depth and increased with the org. matter content in the sediment; FeS₂ and org. S increased with the depth. Exceptions to this rule were ascribed to local migration of FeS and to possible accelerated rates of FeS formation. The rule supports the postulate that H₂S, from the reduction of sulfates by bacteria, forms a hydrous FeS which then combines with elemental S to form FeS₂ (pyrite and marcasite) and org. S.

I. Benoit

Ostrooumov, E. A.

USSR/ Minerals - Geochemistry

Card 1/1 Pub. 22 - 32/49

Authors : Ostrooumov, E. A.

Title : Iron in the bottom deposits of the Okhotsk Sea

Periodical : Dok. AN SSSR 102/1, 129-132, May 1, 1955

Abstract : Geological-biochemical data are presented on the amounts and chemical composition of iron extracted from bottom deposits of the Okhotsk Sea. Fifteen references: 13 USSR, 1 USA and 1 Ger. (1924-1953). Diagram.

Institution : Acad. of Sc., USSR, Inst. of Oceanology

Presented by : Academician N. M. Strakhov, February 9, 1955

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7"

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7

MIL'IN, V. I., MIL'YANTSEVA, N. A.

Applicant is of the opinion that he has no knowledge of the responsibility of the
Soviet Union for the shooting down of the American aircraft over Libya.

1.000-29481-29-0000000000000000

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7"

OSTROUMOV, Z. A.

B.C.

18-1-8

Determination of selenium and tellurium in the selenium sludge of sulphuric acid factories. Z. A. Ostrovskiy (Zavod. Lab., 1934, 3, 1068-1069).—3 g. of material are digested with 20 c.c. of conc. HNO_3 , 10 c.c. of conc. H_2SO_4 are added, heating is continued to elimination of SeO_3 . 50 c.c. of H_2O are added, and the solution is filtered from pptd. BiO_3 and $PbSO_4$.

3 vols. of conc. HCl are added to the filtrate + washings, SO_2 is passed to saturation, the solution boiled under reflux with $N_2H_4 \cdot H_2SO_4$, the Se collected, weighed, and ignited, and the wt. of the residue (chiefly SeO_2) subtracted. Te is determined in the filtrate by adding eq. $FeCl_3$ and eq. NH_3 , collecting the ppt. of $Fe(OH)_3$, containing the Te, dissolving it in HCl, and pptg. Te by saturation with SO_2 as for Se. R. T.

ASB-15A METALLURGICAL LITERATURE CLASSIFICATION

18-1-8
18-1-8
18-1-8

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7

MINKIN, V.I., GUSKOV, M.I., OSTREROV, O.A.; OSTROUMOV, Yu.A.

Electron structure and absorption spectra of salicylanilides
and its derivatives. Opt. i spektro. 18 no.4:571-578 Ap '65.
(MIRA 18:8)

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7"

OSTROUMOV, Z. A.

USSR/Chemistry - Geochemistry

Card 1/1

Author : Ostromov, Z. A.

Title : Manganese in the bottom deposits of the Okhotsk Sea

Periodical : Dokl. AN SSSR, 97, Ed. 2, 285 - 288, July 1954

Abstract : In comparison with manganese deposits of other Soviet seas, the deposits of the Okhotsk Sea are considered to be the highest. Maximum manganese deposits found in the White Sea were 2.84% whereas the manganese deposits of the Okhotsk Sea were 3.07%. The origin of these manganese deposits in the sea are explained. Sixteen references. Table, drawing.

Institution : Acad. of Sc. USSR, Institute of Oceanology

Presented by : Academician N. M. Strakhov, April 20, 1954

OUTLINE OF LITERATURE

Meteorological Abst.
Vol. 4 No. 6
June 1953
Synoptic Analysis and
Forecasting

4.6-69
Ostrovskiy, Z. S., Uprabchennyj sposob opredelenija dat perekroda srednesutochnykh temperatur cherez opredelennye znachenija putem vychislenija. [Simplified method of determining the date of passage of mean daily temperature through definite values by calculation.] Meteorologija i Gidrologija, No. 3:59-60, Nov. 1950. fig., table, 2 refs. DLC—This method requires the solution of a simple geometrical problem: the determination of the magnitude of a single perpendicular in two similar right angle triangles given the magnitude of two other perpendiculars and the magnitude of the sum of the unknown perpendiculars. The equation $b = an/a + a_1$ is derived by the method of proportionality where $a = b + b_1$, the number of days between the fifteenth number of corresponding months. The magnitude b is the number of days which must be added to the mean of the month in order to obtain the date of the passage of the temperature through the given value. Subject Headings: 1. Temperature forecasting. 2. Mathematical techniques.—J.L.D.

(1) Geo

551.509.53:518.2

OSTRUMOV-VASIL'YEVA, V. A.

Ostroumova-vasil'yeva, V. A. "On the problem of the relation of dielectric potential to the metabolic process of living tissues," report I, Ostroumova-vasil'yeva, V. A. "Change of potential on impairment of a skeletal muscle of the frog under the effect of certain chemical and physical factors," Izvestiya Yestestv.-nauch. in-ta pri Molotovskom gos. un-tete im- Gor'kogo, Vol. xii, Issue 8, 1948, p. 351-60 - Bibliog: 12 items

SO: U-2868, Letopis Zhurnal'nykh Statey

Ostroumova A.S.

3(5)

PHASE I BOOK EXPLOITATION

SOV/1638

Vsesoyuznyy nauchno-issledovatel'skiy geologicheskiy institut

Geologicheskoye stroyeniye i perspektivy neftegazonosnosti Zapadno-Sibirs'koy nizmennosti (Geological Structure and the Oil-and Gas-bearing Possibilities of the West Siberian Plain) Moscow, Gosgeotekhizdat, 1958. 390 p. (Series: Its: Trudy) 3,000 copies printed.

Additional Sponsoring Agency: USSR. Ministerstvo geologii i okhrany nedr.

Ed.: N.N. Rostovtsev; Compilers: Z.T. Aleskerova, G.S. Kritsuk, P.P. Li, I.V. Litvinenko, D.V. Osadchaya, A.S. Ostroumova, T.I. Osyko, O.V. Ravdonikas, N.N. Rostovtsev, T.N. Simonenko, M.A. Tolstikhina, B.E. Khesin; Ed. of Publishing House: N.I. Babintsev; Tech. Ed.: K.V. Krynochkina.

PURPOSE: This book is intended for petroleum geologists and economic planners in the oil and gas industry.

Card 1/12

Geological Structure (Cont.)

SOV/1638

COVERAGE: This work, written by several geologists, describes the geology of the West Siberian Plain in relation to its oil and gas potential. It summarizes the results of the initial stage of the second period in the search for oil and gas in Western Siberia and indicates the direction to be taken in changing the approach from a general regional study to a detailed investigation of potential oil and gas areas. The rapidly developing industry, transportation, and agriculture in Siberia are requiring larger and larger quantities of liquid fuels. Only since 1949 has large-scale geological and exploratory drilling along with geophysical, hydrological, and special investigations been carried on. During this comparatively short period a large oilfield was discovered in Berezovo on the Ob' River. It was definitely established that the West Siberian Plain is the repository of some of the world's largest artesian basins with large reserves of thermal (up to 120°C) calcium-chloride and other waters with a 1-60 g. mineralization, saturated with flammable gases, mainly methane. The Introduction contains a detailed listing of the various trusts, research institutes, surveys, and expeditions which have participated in the studies upon which this work is based. In addition, the names of individuals and their special contributions (stratigraphy, luminescent studies,

Card 2/12

Geological Structure (Cont.)

SOV/1638

thermal studies in wells, surveying, etc.) is provided. Some 200 personalities are listed. There are 27 tables, the last of which on the composition of underground waters of the West Siberian Plain, extends for 85 pages. There are 336 references, of which 332 are Soviet, 2 German, 1 English, and 1 French.

TABLE OF CONTENTS:

Foreword (N.N. Rostovtsev)	5
Ch. I. Introduction (N.N. Rostovtsev)	5
Ch. II. Stratigraphy	11
1. Brief history of the studies made on the sedimentary deposits in the folded basement of the plain. D.V. Osadchaya, and N.N. Rostovtsev	11

Card 3/12

Geological Structure (Cont.)

SOV/1638

2. Sediments in the plain's folded basement 12
The probable pre-Cambrian beds 12
Berezovskaya basic borehole 1-R. P.F. Li and
A.S. Ostroumova. 14
Lower Paleozoic (?)
Yakovlevskaya borehole 3-R. Z.T. Aleskerova and
T.I. Osyko 14
Middle Paleozoic 17
Zavodoukovskaya borehole 3-R. D.V. Osadchaya (based
on the materials of P.F. Li). Kolpashevskaya basic
borehole. M.A. Tolstikhina. Barabinskaya basic bore-
hole. T.I. Osyko. Mal'tsevskaya borehole 2-R. D.V.
Osadchaya (based on materials of P.F. Li, and T.V. Dolinina).
Upper Paleozoic
Boreholes of the Ryavkinskaya area (1-R, 3-R, 4-R, 5-R).
Z.T. Aleskerova, Luchinskaya borehole 2-R.
D.V. Osadchaya (based on the materials of P.F. Li and
T.V. Dolinina), Lugovskaya and Uteshevskaya boreholes.
D.V. Osadchaya (based on the materials of P.F. Li and
T.V. Dolinina). Derbyshinskaya borehole. D.V. Osadchaya
(based on A.V. Khabakov's materials). Boreholes of the

Card 4/12

Geological Structure (Cont.)

SOV/1638

Vikulovskaya area (1-R and 2-R). Z.T. Aleskerova and T.I. Osyko. Leushinskaya basic borehole. P.F. Li. and A.S. Ostroumova. Boreholes (1-R and 2-R) of the Tebisskaya area. A.T. Aleskerova. Boreholes (1-R, 2-R, 3-R, 4-R, 5-R, 6-R) of the Yakovlevskaya area. Z.T. Aleskerova, and T.I. Osyko. Slavgorodskaya basic borehole. T.I. Osyko

Paleozoic (less definite)

23

Boreholes (1-R, 2-R, 3-R, 4-R, 5-R) A.T. Aleskerova, and A.S. Ostroumova. Boreholes (1-R, 2-R, 3-R) of the Tatarskaya area. Z.T. Aleskerova and A.S. Ostroumova. Bol'sherechenskaya basic borehole. A.T. Aleskerova (based on the material of Zapsibneftegeologiya Trust). Luchinskaya (1-R) borehole. D.V. Osadchaya (based on A.B. Khabakov's materials). Omskaya basic borehole. Z.T. Aleskerova. Boreholes of the Oktyabr'skaya area. Z.T. Aleskerova, and A.S. Ostroumova.

Card 5/12

Geological Structure (Cont.)

SOV/1638

Lower Triassic (?) (Turinskaya series)	26
Tryumenskaya basic borehole. P.F. Li (based on A.B. Khabakov and Ye. Z. Bur'yanova's materials).	
Yarskaya borehole. D.V. Osadchaya (after A.B. Khabkov and S.M. Chikhachev's materials). Kushmurunskiye bore- holes. D.V. Osadchaya (based on A.G. Ber, and V.P. Gorskiy's materials)	
Upper Triassic (?), Rhetian - Lias (?). T.I. Osyko, Chelyabinskaya series	32
Dronovskaya stage. Omskaya stage.	
3. Stratigraphy of the Meso-Cenozoic cover of the plain Introduction. N.N. Rostovtsev	34
Brief description of the series and some controversial problems in the stratigraphy of mesozoic and tertiary formations. N.N. Rostovtsev	34
Lower and Middle Jurassic (Zavodoukovskaya series). T.I. Osyko	36
Lower and Middle Jurassic (Tyumenskaya stage). Lower and Middle Jurassic (Pokrovskaya stage). Middle Jurassic (Tatarskaya stage). Upper Jurassic-Valanginian Hauterivian (Poludinskaya series). T.I. Osyko	44
	50

Card 6/12

Geological Structure (Cont.)

SOV/1638

Callovian-Valanginian (Tebisskaya stage), Callovian
Lower Valanginian (Mar'yanovskaya substage), Valanginian
(Kulomzinskaya substage). Valanginian-Hauterivian (Tarskaya
stage)
Hauterivian-Barremian-Aptian (?) (Sargatskaya series) 57
P.F. Li and M.A. Tolstikhina.
Valanginian (?) Hauterivian-Barremian-Aptian (?)
Kiyalinskaya-Ilekskaya stage) M.A. Tolstikhina.
Hauterivian-Barremian-Aptian (?) (Kiyalinskaya stage).
M.A. Tolstikhina
Hauterivian-Barremian-Aptian (?) Vartovskaya stage)
M.A. Tolstikhina. Hauterivian-Barremian (Leushinskaya
stage) P.F. Li. Barremian-Aptian (?) (Koshayskaya stage)
P.F. Li. Aptian (?) Albian-Cenomanian-Lower Turonian
(Pokurskaya series) Z.T. Aleskerova, and N.N. Rostovtsev 63
Aptian (?) -Albian (Vikulovskaya stage). Albian-Cenomanian-
Lower Turonian (Khanty-Mansiyskaya stage). Lower Turonian
(Uvatskaya stage)

Card 7/12

Geological Structure (Cont.)

SOV/1638

Aptian (?) Albian-Cenomanian-Lower Turonian (Pokurskaya series of the central and the eastern part of the plain) Z.T. Aleskerova, and N.N. Rostovtsev. Aptian (?) - Albian (carboniferous stratum). N.N. Rostovtsev. Aptian-Albian (Kliyskaya stage). N.N. Rostovtsev (after I.V. Lebedev and M.A. Tolstikhina) Cenomanian-Lower Turonian. Amber-bearing stratum. Z.T. Aleskerova. Cenomanian-Turonian (Simonovskaya stage) N.N. Rostovtsev (after A.P. Anan'yeva and M.A. Tolstikhina). Turonian-Senonian (Derbyshinskaya series). Z.T. Aleskerova, T.I. Osyko, N.N. Rostovtsev, M.A. Tolstikhina 72 Turonian (Kuznetsovskaya stage). Z.T. Aleskerova, Upper Turonian (?) Coniacian-Santonian-Campanian (Slavgorodskaya stage). Z.T. Aleskerova. Maestrichtian-(Gan'kinskaya stage). T.I. Osyko. Turonian-Coniacian-Santonian (Kolpashevskaya stratum). N.N. Rostovtsev, and M.A. Tolstikhina. Santonian-Campanian-Maestrichtian-(Narymskaya stratum). N.N. Rostovtsev, and M.A. Tolstikhina: Senonian (Kasskaya stage). M. A. Tolstikhina 82

Card 8/12

Geological Structure (Cont.)

SOV/1638

Danian stage - Paleocene (Talitskaya stage) P.F. Li. Danian
stage - Paleocene (Symskaya stage) M.A. Tolstikhina. Eocene
(Lyulinovorskaya stage). P.F. Li.
Upper Eocene-Lower Oligocene (Tavdinskaya stage). P.F. Li
Middle and Upper Oligocene and Neogene (Irtyshskaya series)
N.N. Rostovtsev
Quaternary formations. N.N. Rostovtsev

94

95

96

Ch. III. Tectonics

1. Brief critical review of the existing views on the structure of the shield's basement and other general problems. N.N. Rostovtsev and T.N. Simonenko 96
2. The structure of the shield's basement in the light of new data. N.N. Rostovtsev and T.N. Simonenko. 108
3. Brief critical review of the existing views on the structure of the Meso-Cenozoic cover of the shield. N.N. Rostovtsev. 127
4. The surface structure of the folded basement and the

Card 9/12

Geological Structure (Cont.)

SOV/1638

sedimentary shield cover. I.V. Litvinenko, N.N. Rostovtsev, and T.N. Simonenko	138
Introduction. N.N. Rostovtsev	138
Structure of the basement surface and the sedimentary cover of the southern part of the shield.	144
N.N. Rostovtsev	
Preliminary data on the structure of the basement surface and the sedimentary cover of the remaining part of the shield. N.N. Rostovtsev and T.N. Simonenko	154
T.N. Simonenko	
Brief description of the local structures of the plain. I.V. Litvinenko, and N.N. Rostovtsev	157
Information on the structure of the uppermost part of the Meso-Cenozoic cover of the shield and the relation of the cover and the basement structure to the topography of the plain. N.N. Rostovtsev	175
Notes on the history of the morphological development of the folded basement of the shield. N.N. Rostovtsev	181

Card 10/12

Geological Structure (Cont.)

SOV/1638

Notes on the character of changes in the thickness of separate series of stages forming the Meso-Cenozoic cover of the shield. N.N. Rostovtsev	183
Some considerations on the origin of the structures of Meso-Cenozoic cover of the shield. N.N. Rostovtsev	186
5. The history of the geological development of the plains. N.N. Rostovtsev. Introduction. N.N. Rostovtsev	191
Brief critical review of contemporary views on the paleogeography of Mesozoic and Tertiary deposits of the plain and some general problems	193
Brief history of the development of the sedimentary shield in the Jurassic, Cretaceous and Tertiary Periods	198
Ch. IV. Materials on the Geothermy and Hydrogeology Mesozoic Formations of the Plains	211
1. Introduction	211
2. Results of geothermal studies in the southern part of the plain	213

Card 11/12

Geological Structure (Cont.)

SOV/1638

3. Underground waters of Mesozoic formations	225
General remarks	226
Characteristics of water-bearing complexes and horizons	227
mesozoic formations	229
4. Hydrochemical zoning of the plains	231
5. Conclusions	240
Ch. V. Oil and Gas-bearing Possibilities of the Plain	240
1. Introduction	242
2. Brief history of the studies made on the oil- and gas-bearing potentialities of Western Siberia	242
3. Evaluation of the oil and gas potentialities of the shield's folded basement	255
4. Evaluation of the oil and gas potentialities of the Meso-Cenozoic cover of the shield	261
5. Conclusion	282
	378

Bibliography

AVAILABLE: Library of Congress
Card 12/12

6/17/59
MM/Jab

Determination of butyric and propionic acids in acetic acid. A. V. Vinogradov and P. A. Ostromyshev. Zhur. Anal. Khim., 1, 88 R. B. 178 (1957). The acid is first accurately neutralized with NaOH and converted into a mixt. of Na soaps; and the acids are then separated by neutralizing with small amounts of HCl. The sept. org. acid is oxid. with Br₂O₂, neutralized again with NaOH in the presence of phenolphthalein and the Na salt is then tested by adding a few drops of a FeCl₃ soln. The butyric acid salt forms immediately a yellow ppt., while salts of propionic and acetic acids produce brownish solns. which give on heating ppts. of yellow and brownish color, resp. The relative amounts of the 3 acids in the mixt. can be evaluated by sepr. from the soln. under test all org. acids and added HCl, with the samples in the reagent bottles. Formic acid cannot be traced by this method. A. V. B.

BC

B-II-1

Determination of propionic and butyric acids in acidic acid. A. V. Vinogradov and E. A. Ostroumova (J. Appl. Chem. Russ., 1968, 9, 578-580).—The acids are converted into Na salts, which are fractionally titrated with 3*N*-HCl to 'points' of Et₂O, the Et₂O extracts are titrated with 3*N*-NaOH (phenolphthalein), and FeCl₃ is added to the aq. layer; when a yellow ppt. indicates PrOO₂Na, a brown solution yielding a yellow ppt. when heated indicates BuOO₂Na, and a brown ppt. ArOH. The approx. contents of the acids are determined by comparison with known solutions. R. T.

ADD-51A METALLURGICAL LITERATURE CLASSIFICATION

SEARCHED		INDEXED		SERIALIZED		FILED	
SEARCHED	INDEXED	SERIALIZED	FILED	SEARCHED	INDEXED	SERIALIZED	FILED
SEARCHED	INDEXED	SERIALIZED	FILED	SEARCHED	INDEXED	SERIALIZED	FILED

ca

11A

The kinetics of serum colloid precipitation by electrolytes at various ages I. B. B. Ostroumova and V. A. Goldenberg. *J. Physiol. (U.S.S.R.)* 21, 263 (1936).
Rapid proof is offered for the view that a correlation exists between the aging of an organism and the lowering of the degree of dispersion of colloids. With the aid of a specially constructed photocell (electro-nephelometer), the coagulation of serum colloids by AlCl_3 was studied. A comparison of the coagulation curves of the serum of young, adult and old dogs shows a decrease of the stability of serum proteins with age. H. Cohen

ASG 114 METALLURICAL LITERATURE CLASSIFICATION

)(3)
AUTHORS:

Vayser, V. L., Ryabov, V. D.,
Ostroumova, A. K.

SOV/20-125-4-29/74

TITLE:

Catalytic Condensation of 9-Methyl-(1,2), (7,8)-Dibenzoxanthene
With Ammonia (Kataliticheskaya kondensatsiya 9-metil-(1,2),
(7,8)-dibenzoksantena s ammiakom)

PERIODICAL:

Doklady Akademii nauk SSSR, 1959, Vol 125, Nr 4, pp 799-800
(USSR)

ABSTRACT:

The authors obtained the substance mentioned in the title (I) by alkylation of β -naphthol with acetylene in an alcoholic solution in the presence of the catalyst $H_3PO_4 \cdot BF_3$ and one gram mercury oxide (Ref 1) (see scheme I). Compound I forms white crystals with a melting temperature of 173° and a boiling point of $268-269/8$ mm, which are well soluble in acetone, acetic acid and ether and soluble on heating in n-heptane and benzene. The interaction between I and ammonia is explained by means of a scheme (Ref 2). A derivative of acridine 9-methyl-(1,2), (7,8)-dibenzo-9,10-dihydroacridine is to be expected as a result. The authors carried out this reaction on a device (Fig 1). Many experiments were necessary for determining the

Card 1/3

Catalytic Condensation of 9-Methyl-(1,2), (7,8)-Di- SOV/20-125-4-29/74
benzoanthene With Ammonia

optimal conditions: temperature - 470°, at which on the one hand no unreacted product (as e. g. at 350-400°) is obtained and on the other hand, however, (I), a not decomposed product (as at 500°). The volume rate of the added benzene solution of (I) amounts to 0.4/hour. The reaction proceeds under milder conditions and yields better results if in 1 or 2 experiments an already used catalyst is taken. In the case of distilling the reaction products in most cases two fractions are obtained. A. 90 - 140°/2 mm which solidifies to a white crystalline mass and B. 240 - 290°/2 mm which becomes a yellow crystalline mass. After a careful fractionation and re-crystallization white crystals with a melting point of 112° and a boiling point of 131-132°/2 mm were produced from fraction A. The authors identified them as β -naphthylamine. By similar operations light-yellow crystals were obtained from fraction B with a melting point of 213° and a boiling point of 251-253°/2 mm. These crystals were identified as 9-methyl-(1,2), (7,8)-dibenzo-(9,10)-dihydroacridine C₂₂H₁₇N. The authors were the first to obtain this substance. There are 1 figure and 1 Soviet reference.

Card 2/3

WIRGUMOVA, A.S.

Hauyne tephrite of the western Baralagez (Armenia).
Trudy VSEGEI 73:11-16 '64. (MIRA 15:9)
(Baralagez Range.. Tephrite)

OSTRUM'COVA, A. S.

Dissertation defended at the Institute of the Geology of Ore Deposits,
Petrography, Mineralogy, and Geochemistry for the academic degree of
Candidate of Geologo-Mineralogical Sciences:

"Tertiary Vulcanism of Western Daralagaz (Armenia)."

Vestnik "Akad Nauk", No. 4, 1963, pp. 119-145

3 (8)

AUTHOR:

Ostromova, A.

DDV/20-127-4-42/EC

TITLE:

On a Recent Find of Leucite Rocks in Transcaucasia

PERIODICAL:

Naukly Akademii Nauk SSSR 1959 Vol 127, Nr 4 pp 87-876 (USSR)

ABSTRACT:

The author found interesting varieties of leucite rocks so far unknown, in the tertiary volcanogenic mass of the Zapadnyy Dzhavakhet (Western Dzhavakhet) near the villages Karachich and Tarkhut (Rayons Vedienskiy and Mikeyanskiy) of the Armenian SSR. Figure 1 shows sites of similar rocks in Guriya. The deposition shapes of the rocks investigated are manifold; there are numerous dikes, little currents and more seldom, volcanic domes and small conglomerations. The discharged leucite rocks lie discordantly on the reef of the Jamidin suite which is pierced by dikes and domes. It is classified either as leucite (Ref 5) or as Oligocene (Ref 9). The leucite rocks mentioned differ to a certain degree with regard to petrography mainly as a result of depositing conditions. Their mineralogical composition and chemistry, however, are rather closely related. The author describes them under the general name of leucite-tephrites despite a certain olivine content (up to 1%) without separating out the basanite. Dimensions. Only one small dome (7 x 5 m) was found (Ref 5).

Card 1/3

On a Recent Find of Leucite Rocks in Transcaspiam

SOV/DO-17-4-47-60

On a recent find, a massive, pale yellowish-banded texture (width of the bands 2-3 mm at a maximum) can be seen. Likewise, the rocks are clean, distinct and set from the others by their black color, an nephrite-like appearance and a shell-like fracture. However, this is a point of some difference; the composition and structure of the varieties went not above the alkali earth, as seen by means of a crossed-polariscope (Table 1). Thus, almost each is seen to consist of a large eye (Table 1). Thus, the second and third variety of leucite-tephrites described above are identical, whereas the first is very different, a certain degree. It consists of fine-grained tephrite, which is a nephrite-like substance, and rarely a greenish-yellow glassy substance overfilled by a granular leucite. The fine-grained varieties scarcely contain other minerals (mainly augite, biotite, plagioclase) but have a finely crystalline glass substance consisting of plagioclase, leucite and nephrite minerals. Table 2 also shows analysis results of a tephrite from Gurjya. Hence it appears that the are very similar to the rocks described here. Thus it can be seen that the tephrites under discussion are not related to the tephrites of the Pamir-Karakorum Chain which are situated much closer but those from Gurjya which are situated at a much greater distance. There are

Card 2/2

On a Recent Find of Leucite Rocks in Transcaucasia SOV/20-127-4-42, '60

1 figure 2 tables, and 10 references, 9 of which are Soviet.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy geologicheskiy institut
(All-Union Scientific Geological Research Institute)

PRESENTED: November 13, 1958 by V. S. Sebolev, Academician

SUBMITTED: November 9, 1958

Card 3/3

OSTKOMMENITZ, G.S.R.

PAGE 1 BOOK EXPLANATION

REV 5/762

Conferencings on magnetic phenomena. Eds. 1960.

University Conference I. Strelitz plasma study Conference. (Volume 1) Superconducting coils and Plasma Physics. Transactions of Conference. 1960. Leningrad. 320. 1962 p. 1,000 copies printed.

Soviet Academy of Sciences Press. Institute of Physics.

Editorial Board: D.A. Pashchenko, Doctor of Physics and Mathematics, Professor; A.I. Vol'fson, Doctor of Technical Sciences, Professor; Doctor of Physics and Mathematics V.D. Vasil'ev, Candidate of Physics and Mathematics; V.G. Vinokurov, Candidate of Physics and Mathematics; Yu.M. Kondratenko, and V.V. Tsvetkov.

Rev. 6. Population: Tech. Ed. A. Chertyshev

This book is intended for specialists working in the field of magnetohydrodynamics and plasma dynamics. Conference transactions contain the transactions of a conference held in Strelitz, October 1960, on problems in applied and theoretical magnetohydrodynamics. The objectives of the conference were to promote the interaction of the basic trends in theoretical and applied magnetohydrodynamics, establish contacts between the physical and applied aspects of magnetohydrodynamics, and to stimulate the research of theoretical scientists in order to assist in the development of applications of magnetohydrodynamics for different areas of the Soviet Union. More than 150 papers from different parts of the Soviet Union took part in the conference, and 14 papers were read. Similar conferences are held regularly in the future. The next such conference is scheduled to be held in Strelitz in June 1960. In this present edition of the conference, most of the papers have been published in English. The book is divided into two parts. The first part deals with problems in theoretical magnetohydrodynamics and plasma theory, while the second part deals with problems in such aspects of the subject as the use of magnetohydrodynamics in astrophysics (D.A. Pashchenko), applications of magnetohydrodynamics and the investigation of energy conversion processes in magnetohydrodynamics (A.I. Vol'fson), properties and characteristics of plasma in magnetic field (G.V. Grishkov), direct and inverse problems of magnetohydrodynamics (V.P. Seregin), and other subjects. The second part consists of 33 articles. Some of these are general discussions of specific problems, including the critical state of potential distributions for investigation of the properties of liquid metals (N.N. Gerasimov), the influence of electromagnetic waves (P.G. Kondratenko), the Institute of Physics of the Academy of Sciences, Leningrad. Several articles are devoted to such topics as electronic cyclotrons, electron-beam accelerators for nuclear materials, and their application to the industrial industry (including atomic and space power-supply systems). References are given at the end of each of the articles.

Editor 1. Magnetic Field and Other Physically Turbulent Flow of Liquid Metals

Chairman: N. N. Gerasimov. Magnetic Field

Chairman: V. P. Seregin. Stresses of Rotating Magnetic Field

Chairman: A. I. Vol'fson. Use of Dispersed Field Pump for Stressed Liquid Metals

Chairman: V. A. Kondratenko. Use of an Arc Heater for Inductive Heating of Metal

Chairman: V. A. Gerasimov. Influence of Power-Supply Systems on Properties of Rotating Magnetic Field for Stressed Metal in Electric Arc Processes

Chairman: V. A. Kondratenko. The Liquid Pump

Chairman: V. A. Kondratenko. Selection of Optimal Parameters of the Dispersed Field Pump for Stressed Liquid Metals

Chairman: V. A. Kondratenko. Use of a Current-Generating Electromagnetic Field

Chairman: V. A. Kondratenko. Current Generating Electromagnetic Field

Chairman: V. A. Kondratenko. Current Generating Electromagnetic Field

Chairman: V. A. Kondratenko.

U.S.A., U.S.S.R., and China, and the U.S. and its allies,
EUROPE, etc.

REVENTION. The U.S. will prevent the Soviet Union from attacking
by acting proactively to defend itself. This will be done by
U.S.-U.S.S.R. military cooperation.

The U.S. will support the U.S.S.R. in its efforts to defend itself
from the Soviet Union's military aggression.

OSTROUMOVA, I.N.

First appearance of hemoglobin in embryos of the
rainbow trout (*Salmo irideus* Gibbons). Dokl. AN SSSR
147 no.1:263-264 N '62. (MIRA 15:11)

1. Gosudarstvennyy nauchno-issledovatel'skiy institut
ozernogo i rechnogo khozyaystva. Predstavлено akademikom
Ye.N. Pavlovskim.
(Hemoglobin) (Trout)
(Embryology--Fishes)

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7

...
O. (DODD), 1. ... , another RICOER, etc. (and) "blood indicators" in the same line
in one sentence of this." (Baldwin, 2/27, 1971) (Exhibit C-7, page 2)
or one telegram given in the same line.
(E., 20 Jan, 1971, 1...)

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7"

PRIVILEYEV, T.I., STREL'CHIKOV, G.V., BRIKIN V.A., P.M.; OSTROUMOV, I.N.;
KOROLEVA, N.V.

Adaptation of fishes to new conditions of the environment. Vop.
ekol. 5:136-131 (62). (MIRA 16:6)

1. Gosudarstvennyy nauchno-issledovatel'skiy institut ozernogo
i rechnogo rybnogo khozyaystva, Leningrad.
(Fishes--Physiology) (Adaptation (Biology))

STREL'TSOVA, S.V., BRIZNOVA, T.N., KUDANOV, I.N., OSTROUMOVA, I.N.

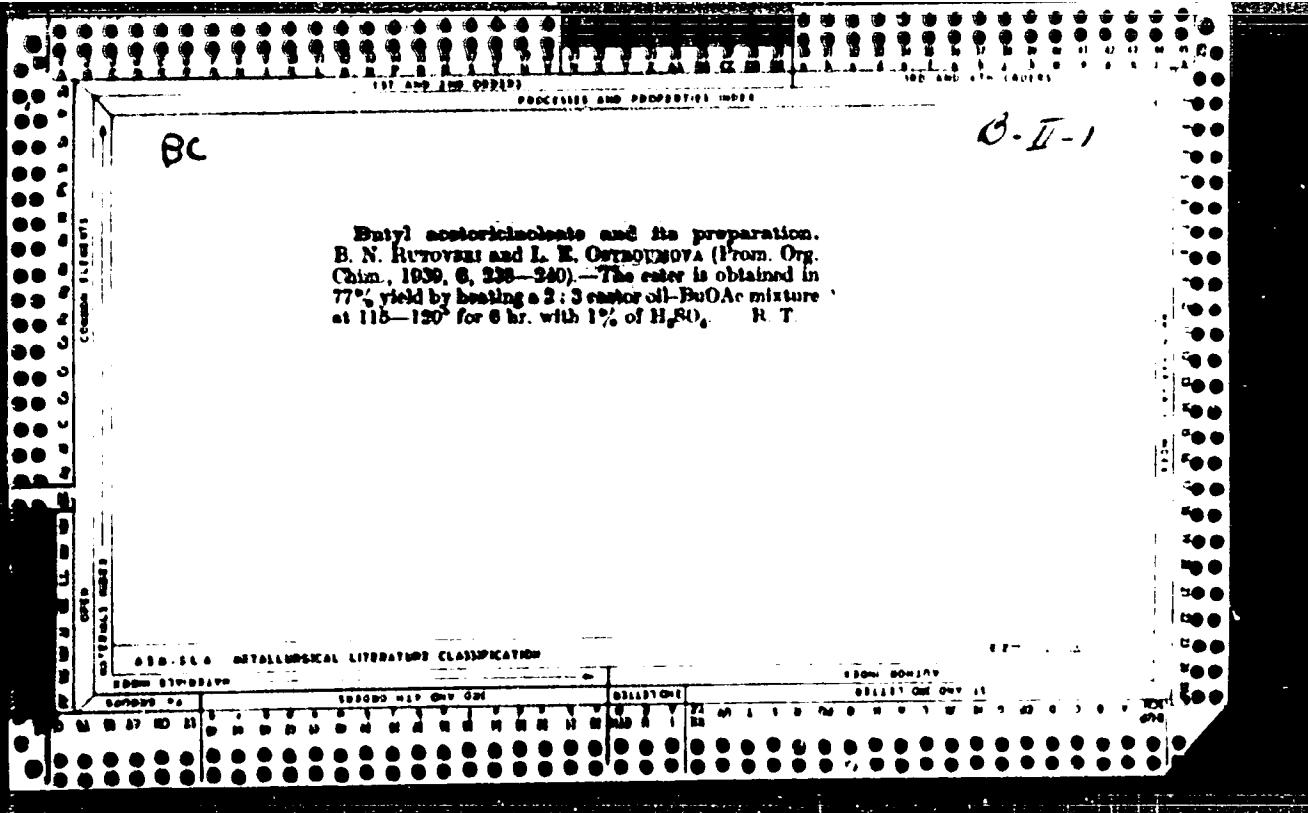
Physiological indices of the same species of fishes in different
geographical locations. Vop. ekol. zool. i zoologicheskogo
'MIRA' 1971.

1. Leningradskiy gosudarstvennyy nauchno-issledovatel'skiy institut
prirody i tekhniki rybnogo khozyaystva.
(Fisika - Physiology)

OSTROUMOVA, I.N.

Blood cells in developing salmon. Trudy sov. ikht.kom. no.8:380-386
'58. (MIRA 11:11)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut ozernogo i rechnogo
rybnogo khozyaystva.
(Baltic Sea-Salmon) (Erythrocytes)



*Co**15*

Preparation of butyl acetoxyisoleate. R. N. Ratury
and E. L. Ostrander, *Jrg. Am. Ind. Chem.* 60, 1809
(1968). Bu-acetoxyisoleate, I, was prepared in
good yields (~) by alkoholysis of castor oil and acylation
of the resulting Bu-isoleate, II, and ~ by interaction
of BuOAc and the oil with the Hg(II) catalyst. 1. Castor
oil and excess of BuOH is refluxed until the alkoholysis is
completed and the reaction mixture is treated with a series of
0.5M NaOH and 1% H₃PO₄ at 10°C for 4 hrs. The
presence of BuOH retards the hydrolysis of II to BuVA.
The reaction mixture is neutralized with 1% NaOH and
distilled with steam or nitrogen to free it from volatile matter
yielding 64.7% I, 82.4% based on castor oil. A mixture
of 100 g. castor oil, 149.2 g. BuOAc and 1% Hg(II), based
on the wt. of oil, is refluxed 6 hrs. at 140°C, yielding
60.5% I, 77.3% based on oil. Redistill of I at a pressure
above 10 mm. results in a higher yield and a
lower ester value. I, η_{D}^{20} = 1.69219, S, has an η_{D}^{20} = 0.34, and η_{D}^{20} = 211.7, η_{D}^{20} = 0.66, η_{D}^{20} = 100.

GEL'FAND, Yu.Ya., vrach OSPICTOVA, L.V., student Ili Karan

Case of metastasis of stomach cancer in the vertebrae form (osteoma).
Sbor. trud. Karak. nos. med. inst. no. 16:376-377 (1960).
(M. A. V. O.)

1. Iz kliniki obshchey kirurgii (zav. - prof. Z.I. Kakhman)
Karskogo medits. tska po instituta.

2000 RELEASE UNDER E.O. 14176

UDC: 678.6

~~UDC: 678.6~~

AVAIL DATE:

1974 - Method of obtaining oligomers which can be polymerized [U.S.S.R.
Sov. 37, 916, 2]

SOURCE: Izobreteniya promyshlennye obraztsy, tovarnyye znaki, no. 9, 1967, 74

TOPIC (ACS): oligomer, polymerization, methacrylic acid

ABSTRACT: An Author Certificate has been issued for a method of obtaining oligomers which can be polymerized. A chloroparaffin or a dehydrochloroparaffin is subjected to interaction at 120-140°C with a potassium or sodium salt of methacrylic acid in a polar initiating solvent, such as a dimethylformamide. [Translation] [NT]

SUB CODE: 07 SUBM DATE: 18Aug64

Cord 1/1

UDC: 678.6

OSTROUMOVA, L.Ye.; AMFITEATROVA, T.A.; SHVAYKOVSKAYA, G.V.; YEGOROVA, L.S.

Thixotropic alkyd resins. Report No.1: Synthesis of polyamides
structuring alkyd resins. Lakokras.mat.i ikh. prim. no.1:23-29
'61. (MIRA 14:4)

(Resins, Synthetic)

(Polyamides)

OSTROUMOVA, L.Ye.; MESHCHERYAKOVA, Z.M.; BONDAREVSKAYA, I.I.

Destructive alkylation of phenol by diisobutylene. Lakokras.mat.
i ikh prim. no.2:16-20 '60. (MIA 14,4)
(Phenol) (Diisobutylene)

15(3)

SOV/DS-11-3-1, 1.

AUTHORS: Meshchervakova, Z.M., Candidate of Technical Sciences, Leningrad, USSR
TITLE: Resins Based on Alkylophenols for the Varnish and Paint Industry
PERIODICAL: Khimicheskaya nauka i promyshlennost', 1980, Vol. 4, No. 3
pp 346-350 (USSR)

ABSTRACT: In the varnish and paint industry phenol-formaldehyde resins are diluted with oils and alkyd resins to obtain hard, resistant and elastic coatings. The assortment of industrial alkylphenol-formaldehyde resins is relatively small. In most cases phenol, more rarely cresol and other phenols, are alkylated. Alkylating agents are halide-alkyls, alkyls and olefines. Catalysts are H₂O₂, CH₃COOH, ZnCl₂, MgCl₂, iron-oxides, resins, etc [Ref. 1]. In the USSR cresol, n-tertiary-butylphenol, mixtures of amyl- and ethyl-phenoles, etc, are produced [Ref. 3, 4]. Alkylphenol-formaldehyde resins are prepared from alkylphenols like phenol-formaldehyde condensates from usual phenol [Ref. 6 - 8]. Resin resins are of greatest interest to the varnish and paint industry. The production of these resins consists of three stages: the addition of formaldehyde to alkylphenols, the condensation of the formed methyl derivatives to resin-like products and the hardening of the obtained

and 1/3

SCN/63-4-3-9/4.

Resins Based on Alkylphenols for the Varnish and Paint Industry

condensates. The activity of phenol is increased by o- and n-nitro ring groups in m-position to the phenolic hydroxyl group. An increase of the basicity of the catalyst raises the amount of polyalcohols and unreacted phenol in the reaction mixture. Thermal hardening is accompanied by the formation of simple ester bonds. At temperatures of 160 - 280 ° the formation of quinone-methides and other complex reactions take place. Resins obtained in the presence of ammonia have a higher molecular weight and melting point than those produced in the presence of an alkaline catalyst. On the base of butylphenol-formaldehyde resin and Al-powder high-quality varnishes and paints are prepared the cost of which is similar to those based on alkyd resins [Ref 2]. Amylphenol-formaldehyde resins with tung oil are used in the production of universal atmosphere-resistant primers [Ref 2]. Alkylphenol resins are also employed for improving other types of varnish raw material [Ref 3]. They are combined with resins based on the polymerization of unsaturated hydrocarbons. The obtained products are soluble in hydrocarbons and resistant to sea water, atmospheric influences and light [Ref 4]. Alkylphenol-formaldehyde resins are employed in the synthesis of polyepoxide resins which have good adhesive properties. The Soviet assor-

Card 2/3

SOV/63-4-3-3.11

Resins Based on Alkylophenols for the Varnish and Paint Industry

ment of phenol resins must be enlarged and the formation of alkyl-phenol resins must be studied. There are 43 references, 8 of which are Soviet, 16 English, 10 German, 7 American, 2 Japanese and 1 Australian.

Card 3/3

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7

Turn 280 v97
1-25-54
Organic Chemistry

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7"

SHCHITAKOV, V. N.; TROFIMOV, V. V.; KUTUZOV, V. I.; ~~REDACTED~~

Acyliks

Synthesis of Acyliks. Part 1. Preparation of Acylidene Acetone
(Acylide). Izv. Akad. Nauk SSSR, Ser. Khim., No. 10, p. 2251, 1958.

9. Monthly List of Russian Acquisitions. Bureau of Intelligence and Research, CIA, Washington, D.C.

SHOSTAKOVSKIY, M.F.; GERSHTEYN, N.A.1 RASKIN, Ya.B.; OSTROUMOVA, L.Ye.
α-Butoxyethyl methacrylate. akad. Nauk S.S.S.R., Inst. Org. Khim., Sintez
Org. Soedinenii, Sbornik 2, 22-4 '52. (MLRA 6:5)
(CA 48 no.2:580 '54.)

KUD'YINA, E.E.; GUTENBERG, N.G.; MAMONOV, V.V.; ZHURAVLEV, V.A.

Thiazolidine-4-thione derivatives. Part I. Synthesis of
3-alkyl-2-thiazolidinones. In: R. et. al., *Voprosy Khimii*,
N. 1, 1970.

I. V. Kud'yinyy na znanie nauchno-tekhnicheskogo komiteta SSSR
Institut imeni G. Ordzhonikidze.

KUZ'MINA, K.K.; OSTROUMOVA, N.G.; MARKOVA, Yu.V.; SHCHUKINA, M.N.

Thiazoline and thiazolidine series. Part 1: Alkylation
of 2-aminothiazoline. Zhur.ob.khim. 32 no.10:3215-3219
O '62. (MIRA 15:11)

1. Vsesoyuznyy nauchno-issledovatel'skiy khimiko-
farmatsevticheskiy institut imeni S. Ordzhonikidze.
(Thiazoline) (Alkylation)

KUZ'MINA, K.K.; OSTROUMOVA, N.G.; MARKOVA, Yu.V.; SHCHUKINA, M.N.

Thiazoline and thiazolidine series. Part 2: Acylation
of 2-aminothiazoline and the reduction of acyl derivatives.
Zhur. ob.khim. 32 no.10:3390-3393 O '62. (MIRA 15:11)

1. Vsesoyuznyy nauchno-issledovatel'skiy khimiko-
farmatsevticheskiy institut imeni S. Ordzhonikidze.
(Thiazoline) (Acylation)

As declassified 4/10/01

Chemical
3

~~Electrochemical deposition of gold coatings of increased hardness. II. N. P. Fedot'ev, P. M. Vaynshteyn, and N. M. Ostromova. Izdat. Leningrad. Tekhnich. Sistem. 1955.~~

~~10-12 g/l. Co to Au cyanide electrolyte increased the microhardness of the deposit by 80%, its wear resistance by 100%, and the rate of deposition of Au by 100%. The deposit contained no Co and its color was not affected. X-ray analysis showed that the crystal grain size was reduced from $\sim 10^{-4}$ to $\sim 10^{-5}$ cm. This accounted for the improved phys. qualities of the deposit. According to the suggested mechanism, the complex Co ions, adsorbed on electrodeposition, were only deformed by the elec. field, forming dipoles. This hindered the discharge of Au ions on preferred points and led to the formation of new cryst. centers and, hence, to the reduced size of the crystals. E. M. R.~~

PM

Ostrov'mova, N.M.

Chem Hard electropolished gold. N. P. Fedorov, N. M. Ostrov'mova, and P. M. Vyacheslavov. J. Appl. Chem. U.S.S.R. 29, 637-9 (1956) (English translation).—See C.A. 50, 14409. 3
B.M.R.

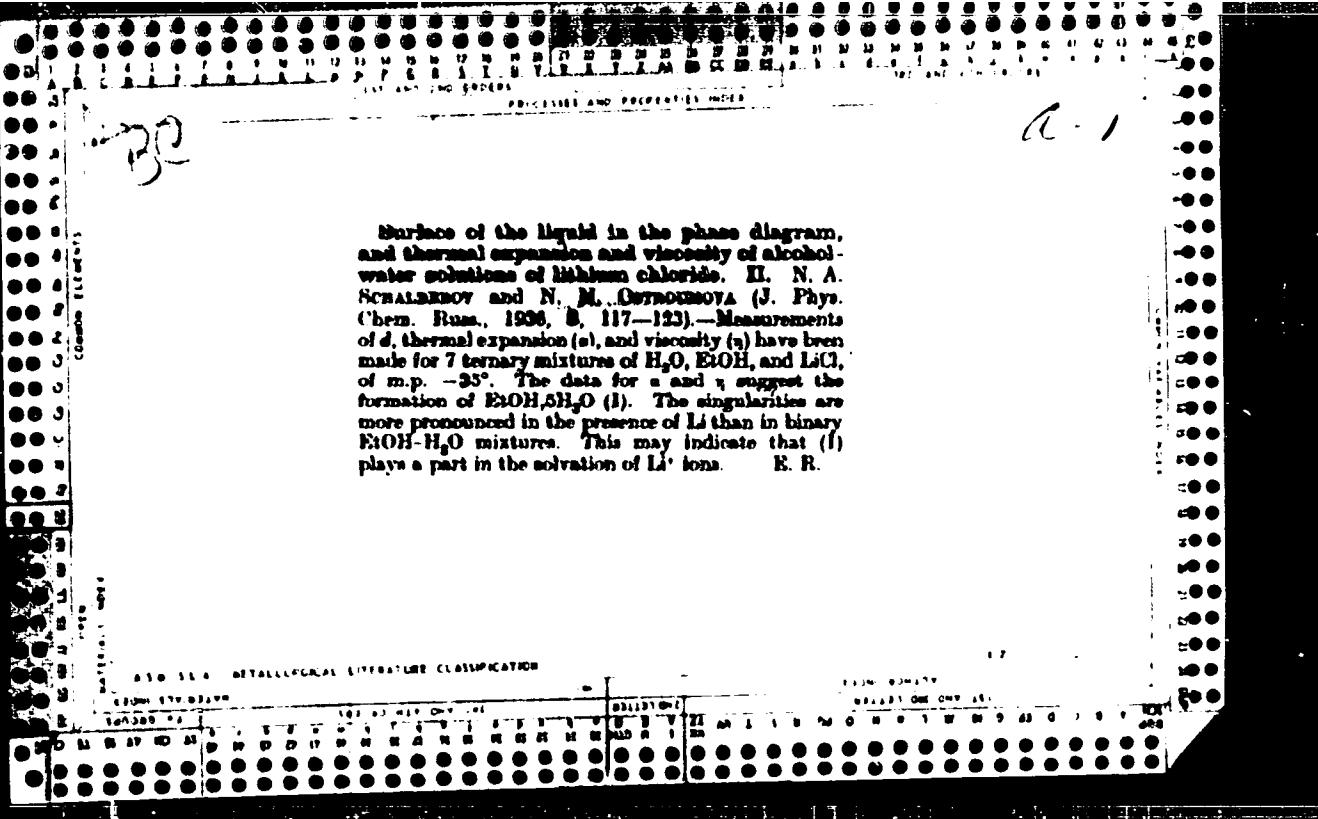
FEDOT'YEV, N.P.; VECHESLAVOV, P.M.; OSTROUMOVA, N.M.; ORILIKHES, S.Ya.

Increasing the durability of gold and silver plated coatings.
Leg.prom. 17 no.3:43-44 Mr 57. (MLRA 10:4)
(Gold plating) (Silver plating)

OSTROUHOVA, N. M.

Battelle Technical Review
July, 1954
Electrochemistry and Electroprocesses

3
6521* Electrochemical Deposition of Very Hard Gold
Const. (Russian.) N. L. Fedotov, N. M. Ostroumova, and
P. M. Vlachislavov, *Journal Prikladnoi Khimii*, v. 27, no. 1,
Jan. 1954, p. 43-50.
Optimum conditions of deposition in Au-Ni baths. Precipitates
tested for microhardness and wear resistance. Tables, graphs.
7 ref



OSTROUMOVA, N. M.

✓ Hardening of gold plating. N. P. Fedot'ev, N. M. Ostroumova
and P. M. Vacharov (Zh. Fiz. Khim., 1953, 27, 450). Co
introduction of Co additives into gold electrolytes, micro-hardening
of gold plate increased 80% (owing to decreased grain size of gold
deposits), wear-resistance increased threefold and precipitation of
gold deposits increased twofold. Micro-hardening only increased
if the Co content was 2-14 g/l. (the greatest degree of hardness
being obtained with c.d. = 2 amp./sq. dm.). Au plate from
electrolytes containing 12 g/l. of Co, c.d. 0.14 and 2.06 amp./sq. dm.
contained no Co.

A. L. B.

PG MT

PEDOT'YEV, N.P.; OSTROUMOVA, N.M.; VYACHESLAVOV, P.N.

Solid gold plating. Zhur. prikl. khim. 29 no. 4:489-492 Ap '56.

(MLRA 9:11)

1. Kafedra elektrokhimii Leningradskogo tekhnologicheskogo instituta
imeni Lensoveta i TSentral'naya nauchno-issledovatel'skaya laboratoriya
kamney samotsvetov.
(Gold plating)

DSTROUMOVKA, N.M.

USSR.

*Electrodeposition of gold plate of superior hardness.
N.P. Fedotov, N.M. Chirikova and P.M. Vachas.
U.S. Pat. Appl. No. 2,713,354 (1954) Engl.
lavor, — See C.A. 48, 63804. H.L.H.*

Ostroumova, N. M.

*Electrodeposition of Gold Coatings of Increased Hardness:
N. P. Fedotov, N. M. Ostroumova, and P. M. Vysotskayev
Zhurnal Tekhnicheskoy Khimii, No. 2, 1955, p. 11; 43-00. [in Russian]
The microhardness (H) of Au deposits obtained from baths
contg. (g./l.) Au 4, free-KCN 16, K_2CO_3 up to 5, Ni 0.5-4.0
(present as oxyside), was determined. The cathodes were of
polished sheet brass, 10×16 mm., and the anodes of Pt,
electrolyzed 30 min. from each side of the cathode. H increased
on adding of the Ni to the bath, and on increasing the c.d. from
1 to 2 amp./dm.², but at 3 amp./dm.² a further increase was
obtained only at the higher Ni contents. Tests at 42° and
70° C. showed that at the lower temp. H was greater but the
deposit was much darker. Subsequent tests were generally
made at 3 amp./dm.² and 70° C. H remained const. as the
KCN concentration increased from 5.1 to 19.1 g./l., then fell
slightly on further increase to 35.5 g./l.. With a bath contg.
(g./l.) Au 4, Ni 2, KCN 16, increasing the K_2CO_3 content from
4.9 to 103.4 g./l. had little effect on H . Increasing the Au
concentration from 1 to 5 g./l. in a bath contg. 3.55 g. Ni/l.
led to a fall in H from 190 to 162 kg./mm.². The wear-resist-

ance (determined by the number of revolutions of a brass
roller necessary to wear away a 2- μ -thick deposit under a load
of 500 g.) of the Au-Ni deposits was 1.61 times greater than
that of the Ni-free deposits. In a bath contg. (g./l.) Au 4,
KCN 16, the current efficiency fell from 24.9 to 20.7% as
the Ni content increased from 0.5 to 4.0 g./l. Increasing the
Au concentration from 1 to 5% in a bath contg. (g./l.) Ni
3.85, KCN 17.9 increased the current efficiency from 7.7 to
22.1%. It fell from 20.2 to 12.1% as the KCN concentration
increased from 8.8 to 50.8 g./l. in a bath contg. Au 4, Ni 2,
 K_2CO_3 7.5. A change in K_2CO_3 concentration from 4.9 to
103.4 g./l. caused the efficiency to fall from 22.5 to 13.0%
for a bath contg. Au 4, KCN 12.5. Increasing the temp.
from 16° to 70° C. in the case of a bath contg. Au 4, Ni 1.6,
KCN 8.8 increased the efficiency from 16 to 22%, at 2 amp./
dm.². Changing the c.d. from 1 to 3 amp./dm.² had little
effect on efficiency. Cathodic polarization curves were
obtained for various baths. The recommended bath contains
(g./l.) Au 4, Ni 2, free-KCN 16, at 2 amp./dm.² and 70° C.

G. V. E. T.

OSTROUMOVA, N.M.

FEDOT'YEV, N.P.; OSTROUMOVA, N.M.; VYACHESLAVOV, P.M.

Electrochemical deposition of gold coatings of greater hardness.
Zhur.prikl.khim. 27 no.1:41-50 Ja '54. (MLRA ? ?)
(Cold plating)

Country : USSR
Category: Soil Science Mineral Fertilizers

Abs Jour: RZhBiol , No 13, 1958, No 82109

Author : Kochergin A. Ye.; Ostroumov O.A.

Inst : -

Title : The Effect of Organic and Mineral Fertilizers on Soils of
Siberia (Summary of Works of Experimental Institutions of
Siberia on the Study of Fertilization)

Orig Pub: Udobrenye i urozhay, 1957, No 8, 8-1)

Abstract: Results are presented of experiments made by the Siberian Agricultural Institute and other experimental stations showing the high effectiveness in Siberian soils of composts of semi-rotted manure with phosphorite fertilizer and Pe, which was especially prepared and

Card : 1/2

z-18

Information from: ~~Refining Company, Inc., Mr. J. P. L.~~ [redacted]

Author: ~~Mr. J. P. L.~~ [redacted], B.B., ~~Refining Company, Inc.~~ [redacted], [redacted]

Title: The Determination of the Elemental Composition of Crude Oil
by Absorption Separation

Abstract: A new method for determining the elemental composition of crude oil by absorption separation is described.

Abstract: The new method of analysis is proposed to determine the elemental composition, with a grain size of 100 microns, of a crude oil by separating the chemical group components of the oil by absorption separation. The analysis is carried out by dissolving the crude oil in heptane; addition of alkylate (PA) with a grain size of 100 microns to the sample is carried out in the following manner: a mixture of heptane, PA and benzene is added to the sample; benzene; then the sample are extracted in a solvent separator. A mixture of heptane and benzene (50% benzene) are added to the sample after completion of the extraction of the proposed method. The sample is petroleum and 10-400 g from katalin, 100 ml of benzene extractum some drawbacks of the method are mentioned, namely, the presence of intermediate fractions consisting of a mixture of naphthalene and aromatic or carbonyl

[redacted]

DIVISION, [REDACTED] ; COORDINATED WITH THE VARIOUS
DIRECTORATES; AND WITH THE FBI.

IT IS RECOMMENDED THAT THE ATTACHED REPORT BE
(B-1, B-2) (C-1, C-2) (D-1, D-2) (E-1, E-2)

ca
OSTROUMOVA, C. A

16

The application of granulated fertilizers on the chernozem
of Siberia. V. I. Kuchegin and A. Ostromova. Siber
Sci. Research Institute. Inst. Zemnoye. Kharkov. 1956.
Vol. 9. No. 12. № 64 (1956). Fertilizer tests with oil
seed crops show that granulated phosphorus has increased
slightly the P content of the crops.

OSTROWMOVA, O.A.

(1)

10617* (Change of Acidity of Granulated Superphosphate
in Chernozem Soil.) Ismenenie kislotnosti granulirovannogo
superfosfata na chernozemnoi pocheve. O. A. Ostrovskaya.
Zemledelie, v. 2, no. 1, Jan. 1931, p. 68-70.

Negative influence on germination of flax and mustard. Tables.

KOCHERGIN, A.Ye.; OSTROUMOVA, N. A.

Dynamics of ammoniacal and nitrate nitrogen in Irtysh region
Chernozems under spring wheat [with summary in English]. Pochvo-
vedenie no.8:86-93 Ag '57. (MIRA 10:11)

1. Sibirs'kiy ordena Trudovogo Krasnogo Znameni nauchno-issledovatel'skiy
institut zernovogo khozyaystva.
(Irtysh Valley--Chernozem soils) (Nitrification)

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7

(STACOMMUL, J. W.)

1
Page 1 doc 9
1970-1971

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7"

OSTROUMOVA, T.M.

Use of radioiodine for function tests of the thyroid gland in heart diseases. Terap.erkh. 29 no.3:82-85 Mr '57. (MLRA 10:8)

1. Iz fekul'tetskoy kliniki (dir. - deystvitel'nyy cheln AMN SSSR prof. V.N.Vinogradov) I Moskovskogo ordena Lenina meditsinskogo instituta imeni I.M.Sechenova
(IODINE, radioactive,
thyroid funct. test in heart dis. (Rus))
(THYROID GLAND, function tests,
radioiodine test in heart dis. (Rus))
(HEART DISEASE, physiology,
thyroid radioiodine funct. test (Rus))

KIRZNER, N.A.; OSTROUMOVA, T.S.

Using the decapsulation method for the bleaching of barite. Lakoras.
mat. i ikh prim. no.3:23-5 163. (MIKA 16:9)
(Barite) (Bleaching) (Paint)

(7)

The kinetics of the coagulation of serum colloids by electrolytes in individuals of various ages I. I. Goldenberg and V. A. Ostromieva. J. Physiol., S. R. 23, 203-208 (1952). *Chem. Zentralbl.* 1952, I, 4072. With the help of an electrophotometer, curves were obtained showing the relation between the coagulation of the serum colloids and the time at which the electrolytes were added. The curves showed that the serum colloids of older experimental animals dogs were coagulated more rapidly and more completely than those of younger animals. This effect is connected with a reduction of the older charge on the colloid particles. M. G. M. 50

BALEX, A. [Bálek, Alexej]; DANEK, S. [Daněk, Stanislav], inzh.; POFF, A. [Poff, Arthur], inzh.; KALVODA, Ya. [Kalvoda, Jan], doktor; SHMID, V. [Schmid, Josef], inzh.; SHKvor, I. [Škvor, J.], doktor; VAITTS, A. [Waitz, Antonín], inzh.; ROMASHKIN, N.I. [translator]; VEKSHIN, G.K. [translator]; TEACHEVA, T.K. [translator]; OSTROUMOVA, V.S., red.; SEMENOVA, N.Kh., red.; KAPRALOVA, A.A., tekhn.red.

[General inventory of fixed assets in Czechoslovakia] General'naja inventarizatsija osnovnykh fondov v Chechhoslovakii. Moskva, Gos. statist. izd-vo, 1959. 101 p. (MIRA 13:2)
(Czechoslovakia--Inventories)

GERSHENOVICH, A.I.; BALAKIREV, Ye.S.; OSTROUMOVA, V.V.

Continuous method of production of alkyl sulfonates.
Khim.prom. no.10:701-707 O '62. (MIRA 15:12)
(Sulfonic acid)

OSTROUMOVA, Yu. A., kandidat meditsinskikh nauk

Basic trends in the works of V.A.Ravich-Shcherbo. Probl.tub. 34
no.3:3-8 My-Je '56. (MLRA 9:11)

1. Iz vtoroy kafedry tuberkuloza TSentral'nogo instituta usozorenii-
stvovaniya vrachey (dir. V.P.Lebedeva)
(TUBERCULOSIS, prev. and control
contribution of V.A.Ravich-Shcherbo)
(RAVICH-SHCHERBO, VLADIMIR ANTONOVICH, 1890-1955)

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7

OTTER COUNTRY - 100% OWNED BY THE CIA

OTTER COUNTRY - 100% OWNED BY THE CIA

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001238510004-7"

6389. Free Vibrations of a Circular Plate Supported at the Edge with a Unidirectional Discharge at the Support. G. Olszmann. Tech. Phys., U.S.S.R. No. 9, pp. 749-754, 1954. In German.—The free symmetrical vibrations of a circular plate are calculated on the usual (Poisson-Kirchhoff) theory of the bending of thin plates, with the boundary conditions $u = 0$ (where u is the displacement) at the edge, and the bending moment per unit length on a section perpendicular to the radius vector r proportional to $\delta^2/\delta r^2$ at the edge. The solution of the "period" equation by graphical and approximate methods is discussed. The motion is damped, so that the solutions are complex. Asymptotic formulae for the frequency and damping factor for the more rapid oscillations (higher overtones) are given. [See following Abstract.] S. G.

A 5
7

A 53
h

6818. Forced Vibrations of a Circular Plate Supported at the Edge. G. Oymenov. Znaka Phys., U.S.S.R. 8, 7, pp. 689-693, 1954. In Ginzburg's thermo theory (see preceding Abstract) is applied to the forced vibrations of a circular plate under a periodic exciting force concentrated at the center. The equations are set out and general deductions made. If the frictional bonding moment and $\delta \theta/\delta r$ at the edge differ in the phase by 90° , so that there is no loss of energy, then the displacement differs in phase from the exciting force by 90° or 270° ; the nodal lines are sharp, and there is a phase jump of 180° . In other cases, the displacements at different points all have different phases; in addition to the standing wave there is also a progressive wave; and there are no nodal lines. That is the previous case is approached the appearance will probably suggest that of the previous case. S. G.